

Download Name Ionic And Covalent Compounds Guide

Name the second atom and end it with "-ide." Now, simply write the name of the second atom as if it were an anion (negatively-charged ion). In other words, just stick "-ide" on the end. In most cases, you will need to drop a letter or 2 off the end of the name, but this is something that will seem more natural the more you do it.

Ionic And Covalent Bonds ?Lab Report Ionic and Covalent Bonds Lab: Ionic and Covalent Bonds Introduction: The purpose of this experiment was to explore the properties of chemical substances that can be used to identify the types of bonds in a chemical substance using a laboratory procedure The two types of bonds being identified were ionic and covalent.

Writing & Naming Formulas of Ionic & Covalent Compounds ©2006 - Douglas Gilliland The Physical Science Series index 13 COVALENT COMPOUNDS Covalent compounds are made from two non-metals and so don't follow the normal ionic rules. You can tell if a compound is covalent because it has prefixes modifying the